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For isothermal operation, the  
work  $W_S$  is calculated using  
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isothermal  $1 P \frac{W}{M} \log M$   
 $P =$  Substituting the pressure  
ratio into this equation and the  
temperature of methane  
before entering the  
compressor yields:  $S$ ,  
isothermal  $kJ W \dots$

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| <b>Label</b> | <b>Gas</b>            | <b>Partial Pressure (atm) at Gas – Diffusion Layer</b> | <b>Partial Pressure (atm) at Bipolar Plate</b> |
|--------------|-----------------------|--|--|
| <b>A</b>     | <b>CO<sub>2</sub></b> | <b>0.47</b>  | <b>0.01</b>                                    |
| <b>B</b>     | <b>O<sub>2</sub></b>  | <b>0.31</b>  | <b>0.47</b>                                    |
| <b>C</b>     | <b>Gas</b>            | <b>Partial Pressure (atm) at Gas – Diffusion Layer</b> | <b>Partial</b>                                 |

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optimal process for the  
recovery and purification of  
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