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33 seconds 66,613 views How many , grams , of $\text{Ca}(\text{OH})_2$ are needed to react with 41.2 g of H_3PO_4 . The , equation , is $2 \text{H}_3\text{PO}_4 + 3 \text{Ca}(\text{OH})_2 = \text{Ca}_3(\text{PO}_4)_2 + 6 \dots$

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